

Enantioselective Bacterial Biotransformation Routes to *cis*-Diol Metabolites of Monosubstituted Benzenes, Naphthalene and Benzocycloalkenes of Either Absolute Configuration

Christopher C. R. Allen,^a Derek R. Boyd,^{*b} Howard Dalton,^{*a} Narain D. Sharma,^b Ian Brannigan,^b Nuala A. Kerley,^b Gary N. Sheldrake^c and Stephen C. Taylor^d

^a Department of Biological Sciences, University of Warwick, Coventry, UK CV4 7AL

^b School of Chemistry, The Queen's University of Belfast, Belfast, UK BT9 5AG

^c Zeneca Specialities, PO Box 42, Hexagon House, Blackley, Manchester, UK M9 3DA

^d Zeneca Bio Products, PO Box 1, Billingham, Cleveland, UK TS23 1LB

Enzyme-catalysed kinetic resolution and asymmetric dihydroxylation routes to enantiopure *cis*-diol metabolites of arenes and benzocycloalkenes of either absolute configuration have been developed using appropriate strains of the bacterium *Pseudomonas putida*.

Biotransformation of monosubstituted benzenes (e.g. **1**, X = H) and disubstituted benzenes (e.g. **1**, X = F, Cl, Br, Me) in growing cultures of a mutant strain (UV4) of *Pseudomonas putida* has been reported¹⁻³ to yield the corresponding *cis*-dihydrodiols **4** (X = I) and **2, 3** (X = F, Cl, Br, Me). Enantiopure (>98% ee) *cis*-dihydrodiols **4** having a (3*S*)-configuration are the normal metabolites of monosubstituted arenes (fluorobenzene being an exception since the (3*R*) enantiomer is also present²) while both enantiomers (e.g. **2, 3**, X = F, Cl, Br, Me) are produced from 1,4-disubstituted arenes^{1,3} (Table 1). Recent studies have shown that *cis*-dihydrodiol derivatives of iodobenzenes can be chemically modified by substitution of the iodine atom.^{1,4} Thus, catalytic hydrogenolysis (H₂, Pd/C, 1 atm., MeOH containing NaOAc)⁴ of the disubstituted *cis*-dihydrodiols **2, 3** (X = F, Cl, Br, Me) provided a route to both (3*S*)-**4** and (3*R*)-**5** enantiomers of the monosubstituted *cis*-dihydrodiols (X = F, Cl, Br, Me) as shown in Table 1.

Metabolism of the mixture of *cis*-diol enantiomers **4, 5** (X = F, Cl, Br, Me), obtained by the above chemoenzymatic route, in the presence of growing cultures of a wild-type strain of *P. putida* (NCIMB 8859, initially grown with naphthalene as a carbon source), resulted in the selective removal of the (3*S*) enantiomer **4** leaving the (3*R*) enantiomer **5** intact in the culture medium (ca. 30% recovered yield, Scheme 1). Since *P. putida* NCIMB 8859 is a wild type strain (unrelated to the mutant strain UV4) which is presumed to contain a *cis*-dihydrodiol dehydrogenase enzyme, it is reasonable to assume that only the (3*S*) enantiomer **4** is further metabolized *via* the corresponding catechol. This kinetic resolution procedure thus leads to a new enzymatic route to the elusive (3*R*)-*cis*-dihydrodiol enantiomer **5** of fluorobenzene, chlorobenzene, bromobenzene and toluene,

and complements the direct oxidation pathway which produces the normal (3*S*) enantiomer **4** through arene metabolism using *P. putida* UV4.² Isolation of the (3*R*)-*cis*-dihydrodiol enantiomer of toluene **5** (X = Me) and fluorobenzene **5** (X = F) exclusively by the kinetic resolution method provides an alternative to the fractional crystallization procedure previously reported for diols **5, 4** of high enantiopurity [e.g. X = Me (80% ee) or F (88% ee), Table 1], which were obtained by hydrogenolysis of the corresponding diols **3, 2** (X = Me and F)^{2,4} and would be useful for obtaining a wider range of the abnormal (3*R*)-*cis*-dihydrodiol enantiomers.

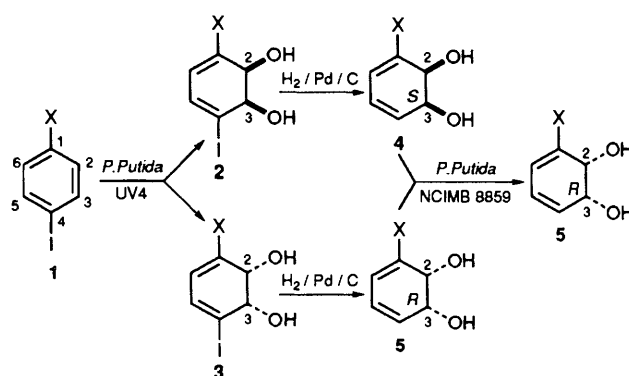
The racemic *cis*-1,2-dihydrodiol derivative of naphthalene **9, 10** was synthesised in four steps, from 1,4-naphthoquinone, by the literature method.⁵ Kinetic resolution of *cis*-diols **9, 10** was examined using two wild-type strains of *P. putida* i.e. NCIMB 11767 (the parent strain isolated using benzene as a carbon source and from which the mutant strain UV4 was derived) and the unrelated strain NCIMB 8859. In both cases the (1*R*:2*S*) enantiomer **10** was removed selectively to yield enantiopure *cis*-diol **9** of (1*S*:2*R*) configuration (ca. 40% recovered yield from the racemate). When naphthalene was metabolized by *P. putida* UV4 it yielded exclusively the (1*R*:2*S*) isomer **10**. Our preliminary study of the enzyme-catalysed kinetic resolution method suggests that it may also be applicable to other members of the polycyclic aromatic and heteroaromatic series where both enantiomers are available.⁶⁻⁸ The (1*S*:2*R*) diol enantiomer **9** had earlier been obtained by kinetic resolution of the racemate using a diol dehydrogenase enzyme from *P. putida* 119.^{5,9}

The value of *P. putida* NCIMB 8859 in the synthesis of enantiopure *cis*-diol metabolites **7** or **8** (*n* = 1-3) was further explored first by using the benzocycloalkenes **6** and then the racemic *cis*-diols **7, 8** as substrates (Scheme 2). Enzyme-catalysed asymmetric dihydroxylation of the alkenes **6** (*n* = 1-3), using a substrate concentration of ca. 0.5-1.0 mg ml⁻¹, yielded exclusively (>98% ee, *n* = 2 or 3) or mainly (ca. 85-90% ee, *n* = 1) the (1*R*:2*S*) enantiomer **8** (ca. 10-30% yield). Our past studies have shown that the *opposite* (1*S*:2*R*)

Table 1 Ee values and absolute configurations of *cis*-dihydrodiols 2-5

Substrate 1	Product diols 2, 3		Substrate diols 4, 5		Product 5
	% ee ^a	Config. ^a	% ee ^b	Config. ^b	
X					Config. ^{c,d}
F	88	2 <i>R</i> :3 <i>S</i>	88	2 <i>R</i> :3 <i>R</i>	2 <i>R</i> :3 <i>R</i> ^e
Cl	15	2 <i>R</i> :3 <i>S</i>	15	2 <i>R</i> :3 <i>R</i>	2 <i>R</i> :3 <i>R</i>
Br	22	2 <i>R</i> :3 <i>S</i>	22	2 <i>R</i> :3 <i>R</i>	2 <i>R</i> :3 <i>R</i>
Me	80	2 <i>S</i> :3 <i>S</i>	80	2 <i>S</i> :3 <i>R</i>	2 <i>S</i> :3 <i>R</i>

^a Major *cis*-dihydrodiol enantiomer **3** from biotransformation of arene **1** in *P. putida* UV4. Ee values and absolute configurations determined by reported methods.⁴ ^b Major *cis*-dihydrodiol enantiomer **5** from catalytic hydrogenolysis of dihydrodiols **2, 3**.⁴ ^c *Cis*-dihydrodiol enantiomer **5** isolated (>98% ee) after kinetic resolution using *P. putida* NCIMB 8859 and biotransformation conditions similar to those reported for *P. putida* UV4.¹¹ ^d Abbreviated to (3*R*) configuration in the text. ^e Fluorobenzene as substrate also yielded the corresponding mixture of *cis*-dihydrodiol enantiomers **4, 5** (X = F)² and the (2*R*:3*R*) enantiomer **5** was again obtained exclusively by an enzyme-catalysed kinetic resolution in the presence of *P. putida* NCIMB 8859.

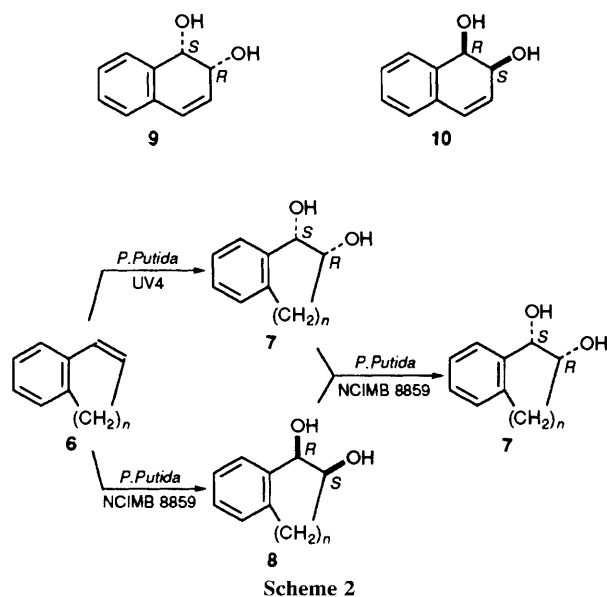


Scheme 1

cis-diol enantiomer **7** was formed exclusively (>98% ee, $n = 2$ or 3) or preferentially (20% ee, $n = 1$) by dioxygenase-catalysed asymmetric dihydroxylation of alkenes **6** ($n = 1-3$) with *P. putida* UV4.^{10,11}

Results obtained using both the UV4 and NCIMB 8859 strains of *P. putida* for dioxygenase-catalysed sulfoxidation have again demonstrated an enantiocomplementary relationship.¹² Thus, in many cases the oxidation products (sulfoxides) were of high enantiopurity but of *opposite* absolute configuration using either strain.

Addition of the racemic *cis*-diols **7**, **8** (obtained by chemical synthesis) as substrates for *P. putida* NCIMB 8859, but using a lower concentration (0.2–0.4 mg ml⁻¹) compared with earlier conditions, resulted in the asymmetric destruction of one enantiomer. The recovered *cis*-diols **7**, **8** ($n = 1-3$) were enriched (35–>98% ee) in the (1*S*:2*R*)-**7** configuration (ca. 35% yield). The relative rates of kinetic resolution decreased with increasing ring size of the benzocycloalkene **6**. The



Scheme 2

Table 2 Ee values and absolute configurations of *cis*-diols **7** and **8**

Alkene 6	Product diol 8 ^a		Substrate diols 7 , 8		Product diol 7 ^b	
	% ee ^c	Config.	<i>n</i>	% ee ^c	<i>t</i> /h ^d	Config.
1	85–90 ^{e,f}	1 <i>R</i> :2 <i>S</i>	1	>98	24	1 <i>S</i> :2 <i>R</i>
2	>98	1 <i>R</i> :2 <i>S</i>	2	91	24	1 <i>S</i> :2 <i>R</i>
3	>98	1 <i>R</i> :2 <i>S</i>	3	35	24	1 <i>S</i> :2 <i>R</i>

^a Major *cis*-diol enantiomer **8** from biotransformation of alkene **6** in *P. putida* NCIMB 8859 (24 h, substrate concentration, 0.5–1.0 mg ml⁻¹).

^b Major *cis*-diol enantiomer **7** isolated after kinetic resolution of diols **7**, **8** using *P. putida* NCIMB 8859 (24 h, substrate concentration, 0.2–0.4 mg ml⁻¹).

^c Ee values were obtained by ¹H-NMR analysis of diMTPA ester derivatives and chiral stationary phase HPLC. ^d Time interval between addition of substrate and product isolation. ^e Higher ee value (90%) after shorter biotransformation times. ^f Accompanied by a minor product resulting from benzylic hydroxylation (inden-1-ol).

recorded % ee values were obtained after a fixed incubation period (24 h) and thus could in principle be >98% under optimal conditions (Table 2).

On the basis of the observations and results contained in this communication we conclude that by using *P. putida* NCIMB 8859: (i) the *cis*-dihydrodiol derivatives of monosubstituted benzenes (and naphthalene) can be obtained exclusively as the abnormal (3*R*)-**5** [and (1*S*:2*R*), **9**] enantiomers by an enzymatic kinetic resolution procedure; (ii) *cis*-diol derivatives of benzocycloalkenes **6** may be obtained with an excess of the (1*R*:2*S*) enantiomer **8** by dioxygenase-catalysed asymmetric dihydroxylation or the (1*S*:2*R*) enantiomer **7** by kinetic resolution of the racemic *cis*-diols **7**, **8**. The enzyme-catalysed routes to enantiopure samples of the (*R*)- and (*S*)-*cis*-diols **4**, **5**, **7**–**10** complement the recent improvements in asymmetric dihydroxylation of alkenes (using OsO₄ and chiral auxiliaries)^{13–15} although the latter chemical route appears to be less stereoselective for *cis*-alkenes and is not applicable to the asymmetric dihydroxylation of arenes.

We thank Mr P. McGeehin for assistance with the hydrogenolysis reactions and the following for financial support: DED/TBNI and The Biotechnology Directorate (to N. D. S.), the SERC/Zeneca CASE award (to C. C. R. A.), DENI/Zeneca CAST award (to N. A. K.) and DENI Quota award (to I. B.).

Received, 6th October 1994; Com. 4/06106C

References

- D. R. Boyd, M. V. Hand, N. D. Sharma, J. Chima, H. Dalton and G. N. Shelldrake, *J. Chem. Soc., Chem. Commun.*, 1991, 1630.
- D. R. Boyd, M. R. J. Dorrity, M. V. Hand, J. F. Malone, N. D. Sharma, H. Dalton, D. J. Gray and G. N. Shelldrake, *J. Am. Chem. Soc.*, 1991, **113**, 666.
- D. R. Boyd, N. D. Sharma, M. V. Hand, M. R. Grocock, N. A. Kerley, H. Dalton, J. Chima and G. N. Shelldrake, *J. Chem. Soc., Chem. Commun.*, 1993, 974.
- D. R. Boyd, N. D. Sharma, S. A. Barr, H. Dalton, J. Chima, G. Whited and R. Seemayer, *J. Am. Chem. Soc.*, 1994, **116**, 1147.
- A. M. Jeffrey, H. J. C. Yeh and D. M. Jerina, *J. Org. Chem.*, 1974, **39**, 1405.
- M. N. Akhtar, D. R. Boyd, N. J. Thompson, M. Koreeda, D. T. Gibson, V. Mahadevan and D. M. Jerina, *J. Chem. Soc., Perkin Trans. 1*, 1975, 2506.
- D. M. Jerina, P. J. van Bladeren, H. Yagi, D. T. Gibson, V. Mahadevan, A. S. Neese, M. Koreeda, N. D. Sharma and D. R. Boyd, *J. Org. Chem.*, 1984, **49**, 3621.
- D. R. Boyd, N. D. Sharma, M. R. J. Dorrity, M. V. Hand, R. A. S. McMordie, J. F. Malone and H. P. Porter, *J. Chem. Soc., Perkin Trans. 1*, 1993, 1065.
- A. M. Jeffrey, H. J. C. Yeh, D. M. Jerina, T. R. Patel, J. F. Davey and D. T. Gibson, *Biochemistry*, 1975, **14**, 575.
- D. R. Boyd, N. D. Sharma, P. J. Stevenson, J. Chima, D. J. Gray and H. Dalton, *Tetrahedron Lett.*, 1991, **32**, 3887.
- D. R. Boyd, M. R. J. Dorrity, J. F. Malone, R. A. S. McMordie, N. D. Sharma, H. Dalton and P. Williams, *J. Chem. Soc., Perkin Trans. 1*, 1990, 489.
- C. R. Allen, D. R. Boyd, H. Dalton, N. D. Sharma, S. A. Haughey, R. A. S. McMordie, B. T. McMurray, G. N. Shelldrake and K. Sproule, *J. Chem. Soc., Chem. Commun.*, 1995, 119.
- B. B. Lohray, *Tetrahedron Asymmetry*, 1992, **3**, 1317.
- S. Hanessian, P. Meffre, M. Girard, S. Beaudoin, J.-V. Sancéau and Y. Bennani, *J. Org. Chem.*, 1993, **58**, 1991.
- K. Morikawa, J. Park, P. G., Andersson, T. Hashiyama and K. B. Sharpless, *J. Am. Chem. Soc.*, 1993, **115**, 8463.